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Many elements exhibit variable valence (particularly transition elements and actinoids). Variation in a period – Increases from 1 to 4 and then decreases from 4 to zero on moving from left to right. Variation in a group – No change in the valency of elements on moving down a group.

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a) All elements (except Se) forms monoxide. b) All elements form dioxide with formula MO_2 , SO_2 is a gas, SeO_2 is volatile solid. While TeO_2 and PoO_2 are non – volatile crystalline solids. c) Ozone: It is unstable and easily decomposes into oxygen. It acts as a strong oxidising agent due to the ease with which it can liberate nascent oxygen.

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